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# Carbon Dioxide As Working Fluid for Power Generation and Refrigeration

Thursday, 10 March 2022 | Technical Topic Webinar

## Presented By

Professor Kandadai Srinivasan – EIT Lecturer

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# Agenda



1	Welcome and Introduction
2	Principle Properties of <b>CO</b> <sub>2</sub>
3	CO <sub>2</sub> as a Working Fluid in Closed Cycle Power Generation
4	<b>CO<sub>2</sub> as a Working Fluid in Vapour Compression</b> Refrigeration
5	Conclusion and Q&A



# Introduction - Presenter





#### Professor Kandadai Srinivasan – EIT Lecturer

- Graduated in mechanical engineering.
- Obtained MS through research from IIT, Madras
- Diploma of Imperial College, London
- PhD from ANU
- Fellow of Engineers Australia, CPEng, NER
- Fellow, ASME
- Near 5 decades of experience in teaching at most renowned institutions across the world, 5 years of industrial experience.
- Specialising in refrigeration, air conditioning, solar energy.
- Numerous consultance projects on energy conservation
- 138 high impact journal publications
- near 4000 citations and h index of 36.

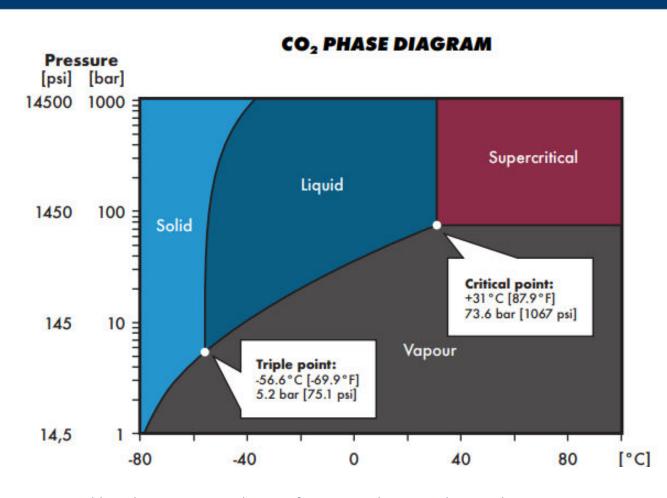
# CO<sub>2</sub> as a working fluid for power generation and refrigeration



- > CO<sub>2</sub> is a natural fluid, non toxic and non flammable
- > A single "green" fluid for both applications
- > Simple system but somewhat higher operating pressures than normal
- > Energy efficient, especially in cold climates
- > Small charge inventories
- > Reduced pipe diameters due to high densities, so low installation cost
- > Technically the best solution available today, given the furore on climate change, global warming
- A lot of research and development is needed for improvements and standardization.

# **Principal Properties of CO<sub>2</sub>**





#### Data from REFPROP

Critical temperature 30.98°C

Critical pressure 73.773 bar

Critical density 467.6 kg/m<sup>3</sup>

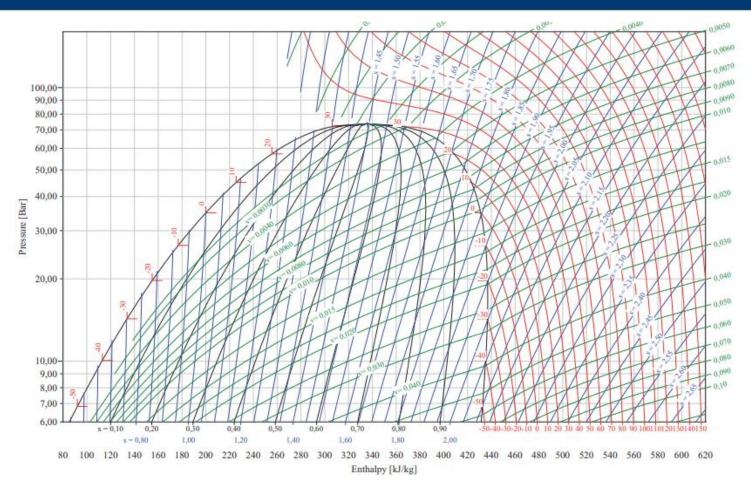
Triple point temperature -56.56°C

Triple point pressure 5.176 bar

From: World Guide to trans-critical CO2 Refrigeration, Shecco Market Development 2020.

# **Pressure-Enthalpy Diagram of CO<sub>2</sub>**





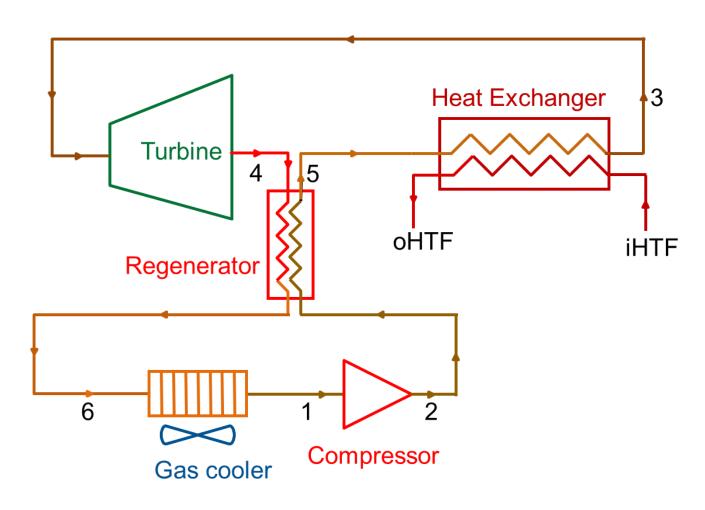
From brochure of Semi-hermetic compressors for R744 (CO2), trans-critical applications of Frascold



# CO<sub>2</sub> as a Working Fluid in Closed Cycle Power Generation

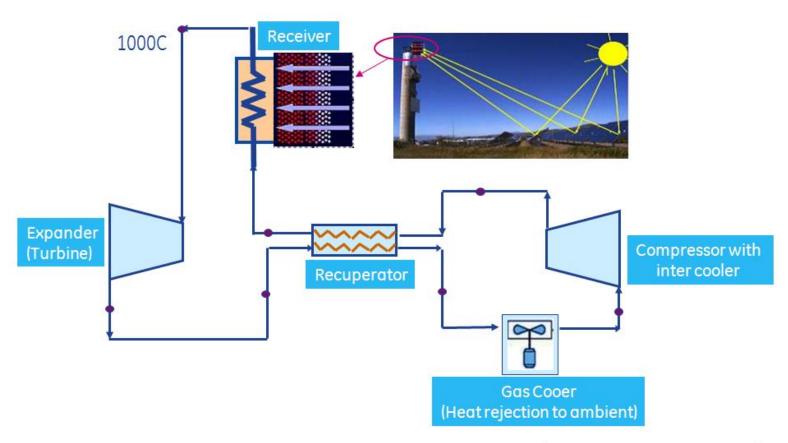
# Simple Brayton Cycle with Regenerator



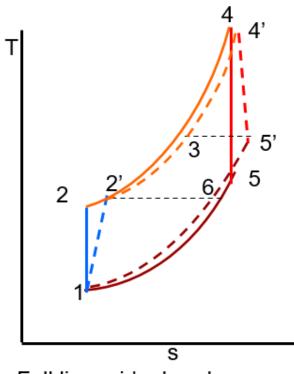




#### **Closed loop CO<sub>2</sub> Brayton Cycle**



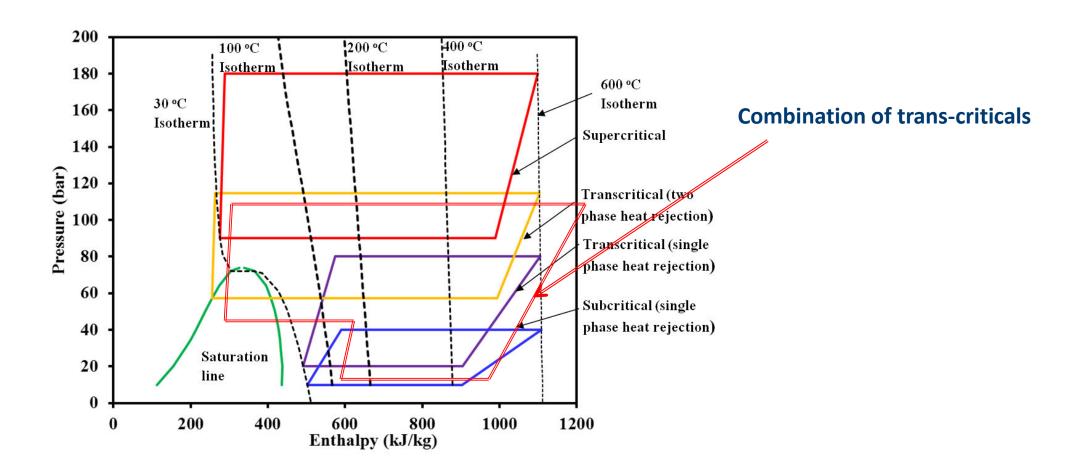
#### T-s diagram of Brayton Cycle



Full line – ideal cycle Dashed line- actual cycle

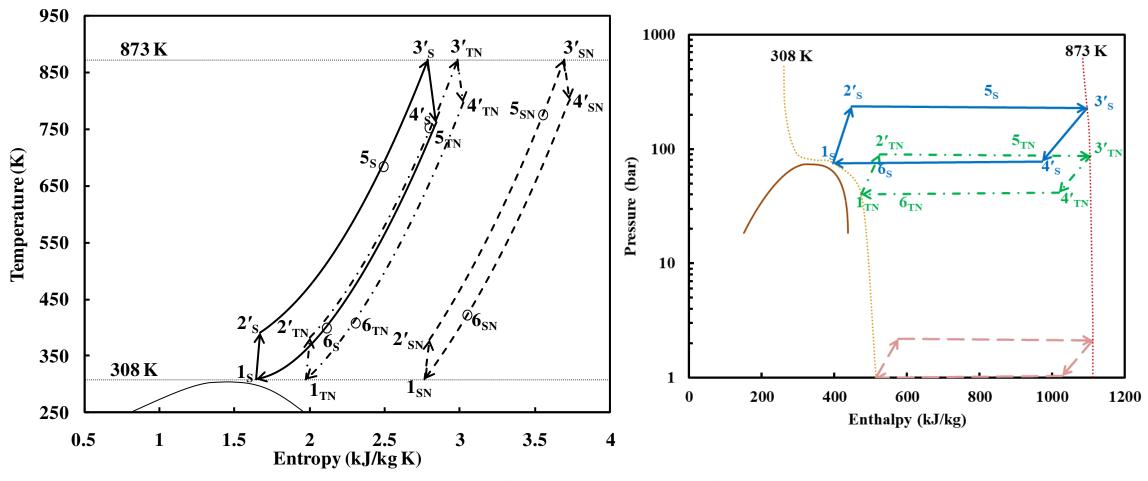
# Options available with CO<sub>2</sub>



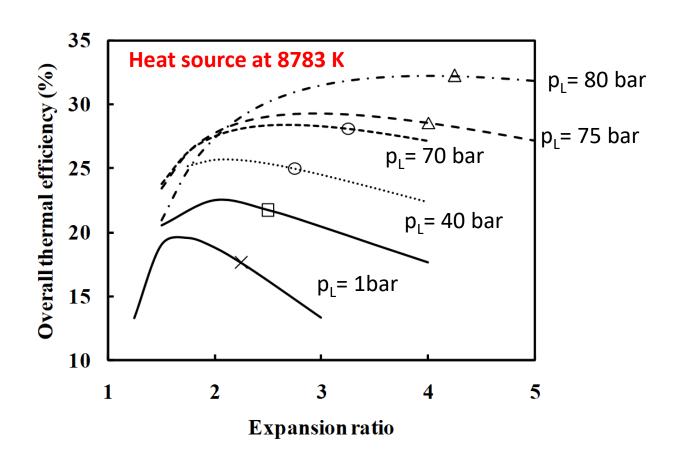


# Simple supercritical CO<sub>2</sub> Brayton Cycle



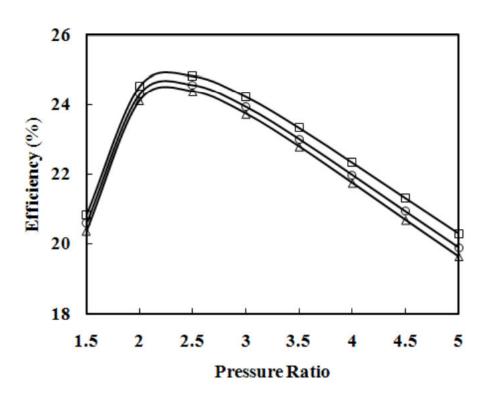




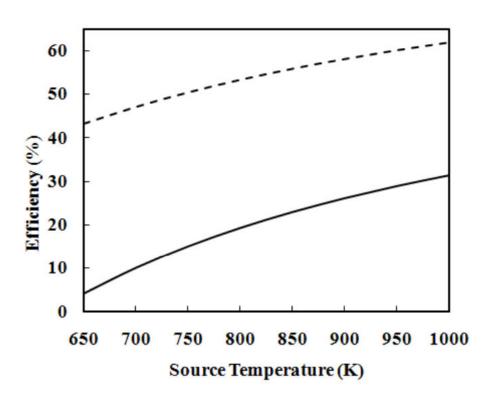


# Some performance indicators (sub-critical)





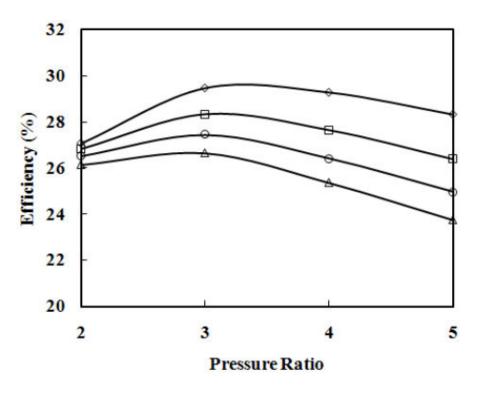
Real cycle efficiency v/s pressure ratio at fixed source temperature of 873 K  $\Delta$  =1bar,  $\circ$  =5bar,  $\Box$  =9bar



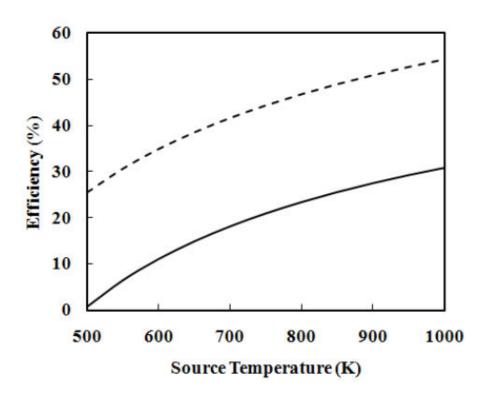
Comparison between ideal and real cycle efficiency for pressure ratio of 2.3 and low side pressure of 1bar at different source temperatures.

# Some performance indicators (trans-critical)





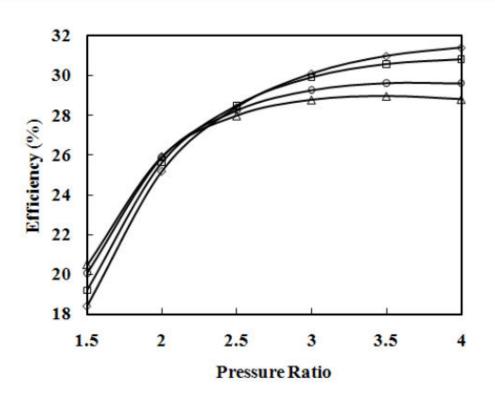
Real cycle efficiency v/s pressure ratio at fixed source temperature of 873 K.  $\Delta$  = 40 bar, 0 = 50 bar,  $\Box$  = 60 bar,  $\Diamond$  = 70 bar



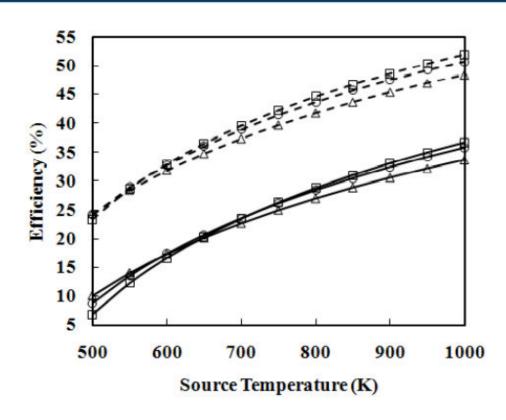
Comparison between ideal and real cycle efficiency for pressure ratio of 2.3 and condenser pressure of 50 bar at different source temperatures

# Some performance indicators (supercritical)





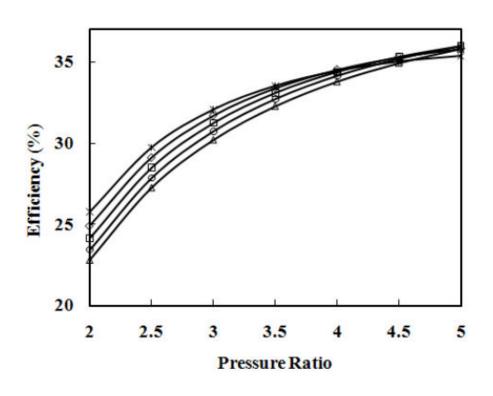
Real cycle efficiency v/s pressure ratio at fixed source temperature of 873 K.  $\Delta$  = 75 bar, 0 = 77 bar,  $\Box$  = 79 bar,  $\Diamond$  = 81 bar

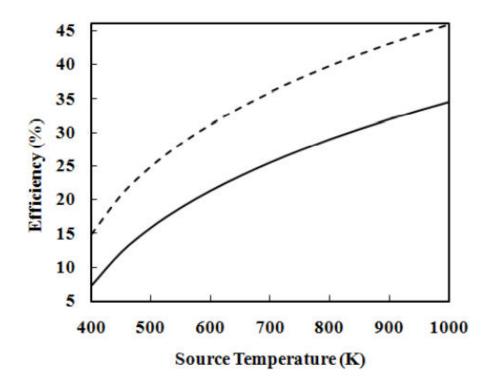


Comparison between ideal and real cycle efficiency for pressure ratio of 2.3 and condenser pressure of 50 bar at different source temperatures

# Some performance indicators (Brayton-Rankine)





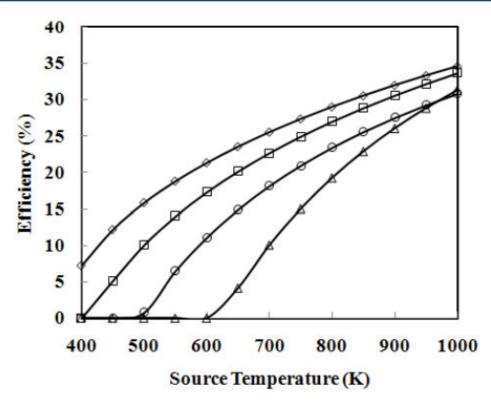


Real cycle efficiency v/s pressure ratio at fixed source temperature of 873 K for different condenser temperature  $\Delta = 280$  K, O = 285 K,  $\square = 290$  K, O = 285 K,  $\square = 290$  K, O = 285 K,  $\square = 290$  K

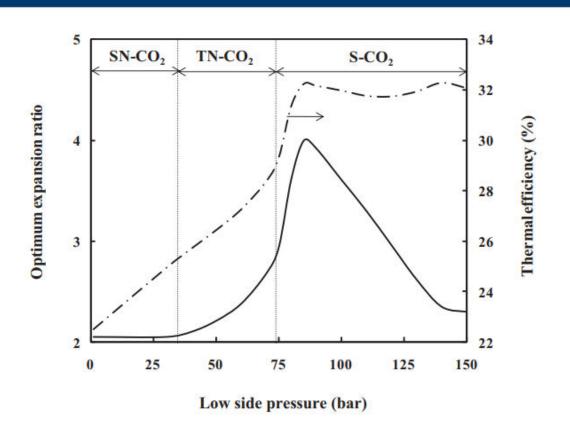
Comparison between ideal and real cycle efficiency for pressure ratio of 2.5 and condenser pressure of 67.13bar at different source temperatures.

## Comparison of all the four possibilities under their optimal conditions





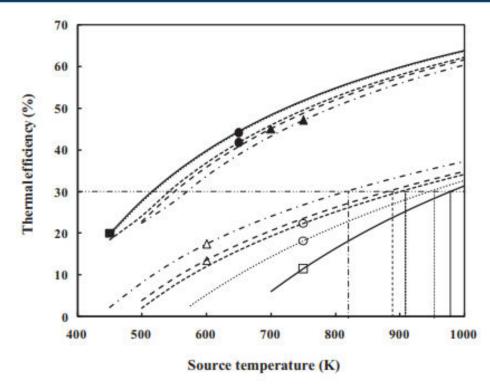
Real cycle efficiency v/s at fixed source temperature of 873 K for the optimized configurations of different possibilities of CO2 cycle  $\Delta$  subcritical,  $\circ$  trans-critical,  $\circ$  trans-critical condensing



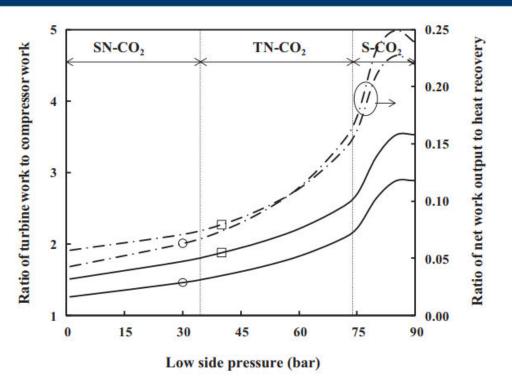
Optimum expansion ratio and corresponding thermal efficiency vs low side pressure for 873 K.

## Further comparison of all the four possibilities under their optimal conditions





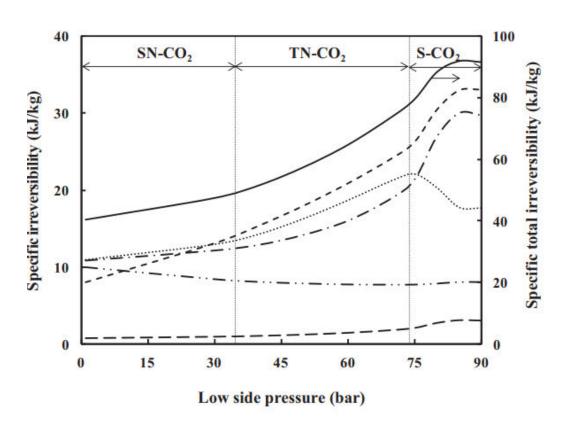
Ideal and real cycle efficiency vs turbine inlet temperature for different low side pressures at their optimum expansion ratios. p1 = 1 bar,  $\cdots$  p1 = 40 bar,  $\cdots$  p1 = 70 bar,  $\cdots$  p1 = 75 bar,  $\cdots$  p1 = 85 bar, SN-CO2 cycle, TN-CO2 cycle, S-CO2 cycle, ideal SN-CO2 cycle, ideal TN-CO2 cycle, and ideal S-CO2 cycle



Ratio of turbine work to compressor work and net work output to heat recovery vs low side pressure at source temperature of 820 and 978K. ratio of turbine work to compressor work (left ordinate),  $-\cdot -\cdot -$  ratio of net work output to heat recovery (right ordinate), T3 = 820K, and T3· = 978K

## Reasons for low efficiencies despite high pressures and temperatures





Component wise and total specific irreversibility vs low side pressure at source temperature of 899K. Sub-systems (left ordinate); boiler,  $\cdots$  compressor, - - - regenerator,  $- \cdot - \cdot -$  turbine,  $\cdots$  gas cooler; system (right ordinate), and total.

- Regenerator and gas cooler are the biggest sources of irreversibility. Both are heat exchangers.
- At low pressures and non condensing subcritical cycles the compressor takes away most of turbine work.
- Even in supercritical operation almost a third of work is taken away by the compressor



#### MAIN DISADVANTAGES

- > The compressor work is a sizeable fraction of expander output
- > kW or even low MW level power generation will be impractical because the mass flow rates are small and so are volumetric flow rates.
- > This causes partial admission around the expander rotor or requires extremely high speeds of rotor (~30000 rpm).
- > High rpm necessitates a gear box if connected to an alternator
- > Reciprocating expanders are impractical because of high power output needs and low isentropic efficiencies.
- > To derive decent work output high-temperature source will be needed (such as concentrated solar, deep earth geothermal) if renewable sources are to be used

#### **ADVANTAGES**

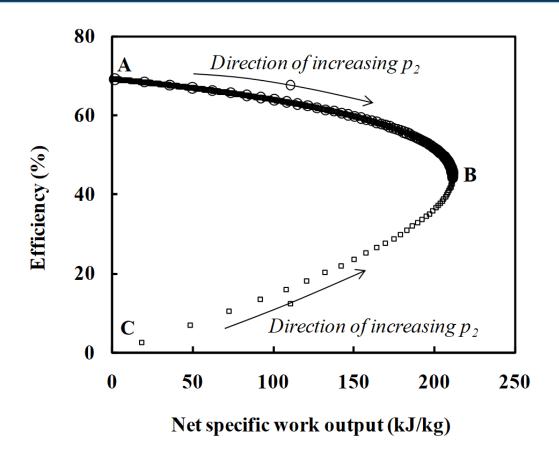
- > Because low specific volumes piping can be smaller and compact than in the case of air
- > Because of excellent heat transfer properties heat exchangers will be compact.
- High temperatures can be obtained just like in gas turbine power plants.
- > No corrosion problems as in gas turbines and moisture is always present in combustion products

All other out put and efficiency improvement processes such as multistage compression, reheat during expansion, regeneration are as applicable to CO2 based Brayton cycles as conventional ones.

Eg: Multistage compression reduces compressor work; Reheat increases work output Both can be used to increase the expansion ratio (as well as compression ratio)

## Performance issues of CO<sub>2</sub> Brayton Cycles



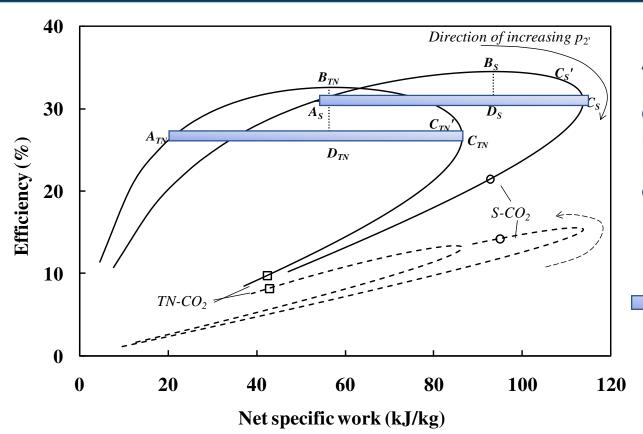


Efficiency vs net specific work for air Brayton cycle with  $p_1$  = 1 bar,  $T_1$  = 308 K,  $T_3$  = 1000 K. Legend:  $\Box$  without regeneration,  $\bigcirc$  with regeneration

- Interestingly, two parameters we are interested are the efficiency and power output. Although, one likes to have both of them to be maximum, they never occur at the same operating conditions.
- Most of academia emphasize on efficiency (or COP in the case of refrigeration)
- Industry is invariably concentrates on maximising power output (or cooling capacity in the case of refrigeration)

## Nature of efficiency-specific work relations in general





A-B efficiency and specific work increase
B - the point of maximum thermal efficiency
C -point of maximum specific work
B-C -the direction of increasing expansion ratio,
Below C - both efficiency and specific work output
decrease, a region is not expected to be of interest.

Zone of regulation without sacrificing efficiency

Efficiency vs net specific work for TN-CO2 and S-CO2 cycle at T3' = 1000 K. Legend: — with regeneration, -- without regeneration,  $\Box$  TN-CO2 cycle (p1= 50 bar), o S-CO2 cycle (p1= 75 bar). Direction of increase in p2' is clockwise for a cycle with regeneration and anti-clockwise when the cycle is without regeneration



- > With receiver temperature in the range of 800-1000 °C, these cycles can have efficiency as high as 50%.
- > In closed loop Brayton cycles are more efficient in multistage compression with intercoolers and regeneration.
- > These cycles do not need a bottoming ORC cycle.
- > Energy storage is possible using a thermal energy storage like pebble bed or silica based rock
- Safe operation from hazardous substance and flammability point of view but need high pressures

#### Heat transfer and pressure drop issues



$$h \propto k\mu^{-0.8} \text{Pr}^{0.33} \, d^{-1.8}$$

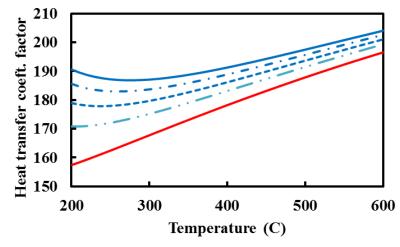
Virtually no difference between

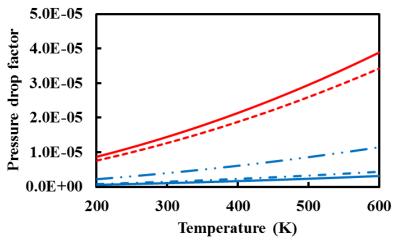
- the two streams for same d at high temperature end.
- 75 and 80 bar warm streams

#### Cooler stream

$$\Delta p \propto \mu^{0.25} \ \rho^{-2} \ d^{-4.75}$$

Low pressure stream has a substantially larger pressure drop, mainly at the point of entry to regenerator for same d.





## **Critical science and engineering challenges**



- > Development of high temperature receivers for Brayton cycle, with  $CO_2$  as the working fluid.
- > High temperature thermal energy storage and low cost reflectors for power tower systems, if solar power is used
- > Development of high efficiency compressors and turbo-expanders
- Biggest liability is the compressor power. If this can be taken out, there will be no limitation of pressure ratios nor other limitations imposed by it.



# **CO<sub>2</sub>** as a Working Fluid in Vapour Compression Refrigeration

# History of CO<sub>2</sub> as a Refrigerant



1850	First patented
1860	First refrigeration system built
1920	CO <sub>2</sub> became more widely used
1950s	The last CO <sub>2</sub> systems were installed in marine applications
1993	The revival of CO <sub>2</sub> refrigeration technology with the first subcritical systems
	being installed again
2000s	First trans-critical system in a supermarket
2008	The introduction of parallel compression and subsequently ejectors led to a much higher adaptability of trans-critical ${\rm CO_2}$ in regions with high
	ambient temperatures

## Refrigerants used for cooling



Basic requirement - non-ozone depleting

- non- greenhouse gas

Pre-Montreal Protocol era - R-22 and R-502

Post Montreal Protocol options - R-507a, 404a, 407c, 410a

(all are blends of zero ODP refrigerants)

Post Kyoto Protocol options

"good old" ammonia

(not widely accepted yet in supermarkets because of smell)

Flammable refrigerants -Propane/Propylene

(not fully accepted because of safety reasons)

Natural refrigerant - CO<sub>2</sub>

High operating pressures (~30 bar in evaporator)

## Why CO<sub>2</sub> is coming back as a refrigerant



Refrigeration and air conditioning is deemed to be responsible for nearly 40% of greenhouse gas emissions.

#### Global Warming Potential (GWP)

It is a measure of how much a given mass of Greenhouse gas is estimated to contribute to global warming. It is a relative scale which compares the gas in question to that of the same mass of Carbon Dioxide.

Refrigerant	R12	R22	R502	R404a	R507	R407c	R134a	R744	R717
GWP	10600	1700	4500	3800	3900	1700	1300	1	0



#### Total Equivalent Warming Impact. (TEWI)

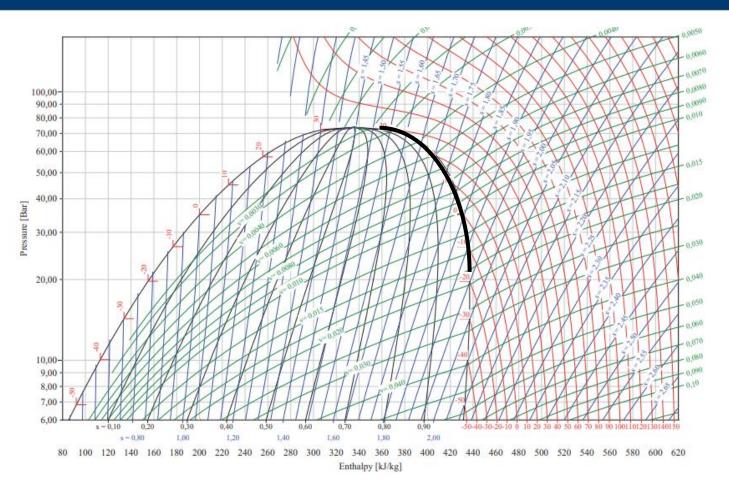
It is the total global warming impact resulting from direct emissions of refrigerant leaking into the atmosphere plus the total indirect global warming impact arising from the indirect emissions as a result of the production of electricity that will be consumed by the plant during operation.

TEWI = GWP Direct + GWP Indirect.

For CO2 direct emission contribution will be nearly zero

## **Pressure-Enthalpy Diagram of CO<sub>2</sub>**





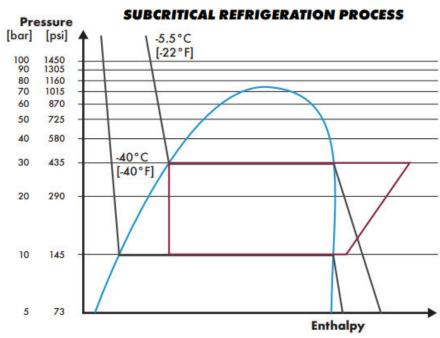
Zone in which most refrigeration systems work.

Note the —ve slope region of saturated vapour curve in p-h plane.

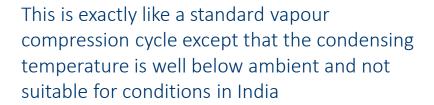
This is a special feature of CO2

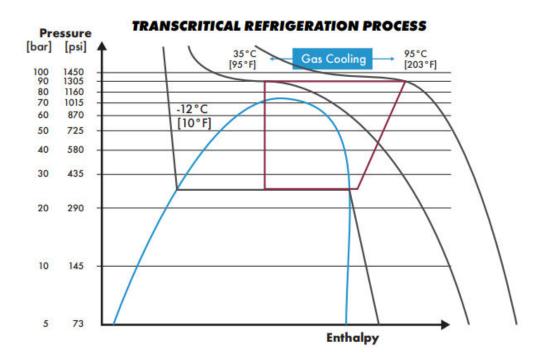
From brochure of Semi-hermetic compressors for R744 (CO2), trans-critical applications of Frascold







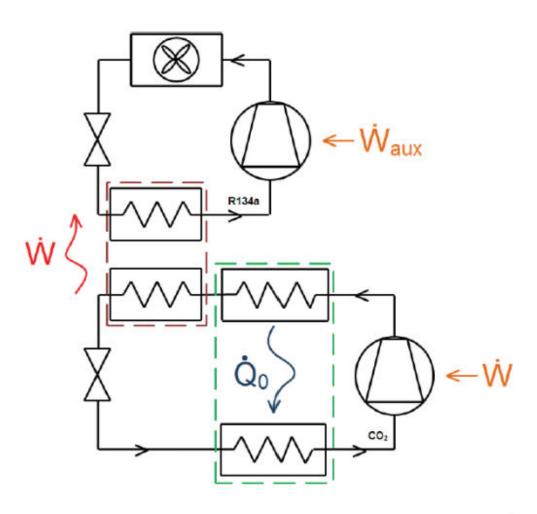




Unlike in standard vapour compression, there is no condenser. It is replaced by a gas cooler. Note that now that it is in supercritical region, gas cooler exit pressure and temperature are not related and each has to be chosen.

### Cascaded CO<sub>2</sub> refrigeration system with upper cycle operated with HFC





- There are still some problems with trans-critical CO2 refrigeration cycles (pressures, controls etc.)
- Subcritical CO2 is more or less standardized
- The MT compressors are replaced by a conventional HFC operated system in a cascaded mode.
- The condenser load of CO2 cycle is dumped on to evaporator of the upper stage

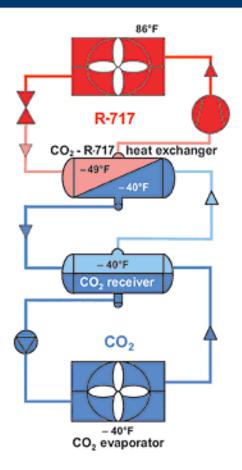
## **Liquid CO<sub>2</sub> Circulation Systems**



- > CO2 is used as a secondary refrigerant (brine).
- > Fully liquid recirculated CO2 systems would need a large volumetric flow rate at about temperatures where refrigeration is required because of its low specific heat (2.23 kJ.kgK). This will necessitate large pumps.
- CO2 must be at a pressure where required subcooling is required.
  Eg: for -10°C refrigeration, CO2 temperature my be allowed to change from -20 to -15°C. At -15°C, saturation pressure is about 23 bar. The liquid CO2 pressure should be ≥25 bar.
- > To reduce volumetric flow rate a small amount of CO2 is allowed to evaporate, because of its large latent heat (~271 kJ/kg)
- > The vapour generated is condensed using the main refrigeration plant that could be operated with HFC's or ammonia.

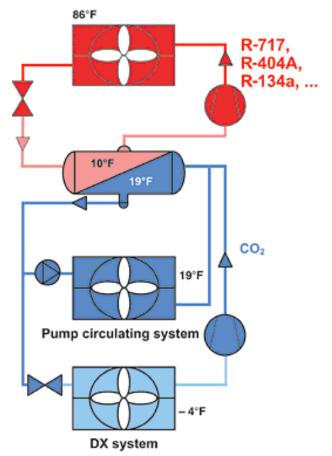
## Some configurations of liquid recirculation systems





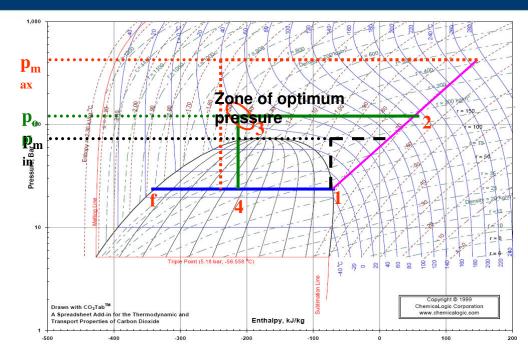
A typical CO2 cascade system with liquid recirculation. CO2 system rejects heat to upper stage cascade system operated with any other refrigerant (ammonia in this figure)

A typical CO2 cascade system with two temperature levels Direct expansion for low temperature and liquid recirculation for medium temperature



## How to determine the optimum gas cooler pressure





We can choose the upper cycle operating pressure (an option we did not have in condensing vapour cycles)

We can maximise

either the cooling capacity (at the cost of very high pressure)

or

COP (with a sacrifice in cooling capacity)

Minimise the entropy generation due to throttling

$$COP = \frac{h_{1} - h_{4}}{h_{2} - h_{1}} \frac{dCOP}{dp_{3}} = 0$$

$$-\frac{\partial h_{3}}{\partial p_{3}}\Big|_{T_{3}} (h_{2} - h_{1}) - \frac{\partial h_{2}}{\partial p_{2}}\Big|_{s_{1}} (h_{1} - h_{4}) = 0$$

$$COP = \frac{h_{1} - h_{4}}{h_{2} - h_{1}} = -\frac{\partial h_{3}}{\partial p_{3}}\Big|_{T_{3}} / \frac{\partial h_{2}}{\partial p_{2}}\Big|_{s_{1}}$$

$$COP = \frac{h_{1} - h_{4}}{h_{2} - h_{1}} \frac{dCOP}{dp_{3}} = 0$$

$$-\frac{\partial h_{3}}{\partial p_{3}}\Big|_{T_{3}} (h_{2} - h_{1}) - \frac{\partial h_{2}}{\partial p_{2}}\Big|_{s_{1}} (h_{1} - h_{4}) = 0$$

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$$COP = \frac{h_{1} - h_{4}}{h_{2} - h_{1}} = \frac{v_{3}(\beta_{3}T_{3} - 1)}{v_{2}}$$

$$COP = \frac{h_{1} - h_{4}}{h_{2} - h_{1}} = \frac{v_{3}(\beta_{3}T_{3} - 1)}{v_{2}}$$

$$F_{3} = \frac{1}{T_{3} - T_{1}}$$

$$s_3$$
- $s_4$  must be minimised  
 $s_4$ =  $s_f$  +( $h_3$ - $h_f$ )/ $T_1$ 

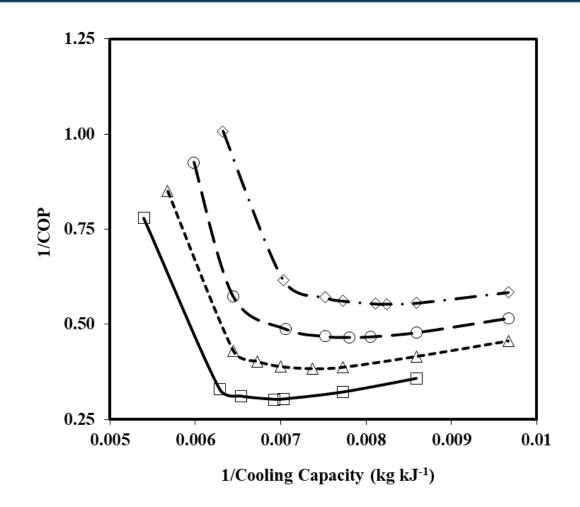
$$\frac{\partial s_3}{\partial p_3}\Big|_{T_3} - \frac{\frac{\partial h_3}{\partial p_3}\Big|_{T_3}}{T_1} = 0$$

$$-\beta_3 T_1 = 1 - \beta_3 T_3$$

p (bar) for -10/40 °C  
min =74.4  
for 
$$Q_{max}$$
=436  
 $\Delta s_{thmin}$  = 110.2  
Ideal COP =2.1



- In any refrigeration system we can have either maximum COP or maximum cooling capacity.
- > They do not occur at the same operating condition.
- Industry looks for maximising cooling capacity. This is because one has already invested in equipment and plant installation. Better to sacrifice some COP to get maximum cooling capacity.
- However, in the case of CO2 it is impossible to maximise cooling capacity due to requirement of very high pressures.



## Positive aspects of CO<sub>2</sub> refrigeration



- > Because of low specific volume of CO2 at operating conditions, the pipe line sizes will be small. This offsets the high pressure liability to an extent because thinner thicknesses may be used.
- > High discharge temperatures from compressor provide an opportunity for heat recovery. This will be advantageous in dairy industry where heating and cooling are required.
- > Because of good heat transfer properties overall size of heat transfer equipment can be reduced
- Because of various configurations available (discussed earlier) enormous flexibility is available to operate it as a stand alone system or in conjunction with other normally used refrigerants

## Problems associated with CO<sub>2</sub> refrigeration



- Pressure ratios  $(p_g/p_e)$  will be large. Also pressure differential in the compressor will be large causing significant leakage from compressed gas to crank case.
- > Isentropic index of CO2 is large
- > Both of them together cause high discharge temperatures (>120°C)
- Polyolester lubricating oils have to be used which are highly hygroscopic. They cannot be exposed to ambient humidity
- > Because of high discharge temperatures oil cooler must be a part of the system. Also oil management is an issue
- Although high discharge temperatures provide an opportunity for heat recovery, it should not be a criterion for choosing CO2.
- > Trans-critical gas cooler design correlations are still evolving and design procedures need standardization.

  Thermophysical properties change over a wide range in the critical region.
- > Supercritical CO2 is difficult to store. CO2 must be expanded and stored in liquid form. Adequate additional refrigeration is needed to cool the receiver.
- > The quality of CO2 admitted to evaporator will be quite large, hence only a small part of latent heat can be used.

## Two stage compression

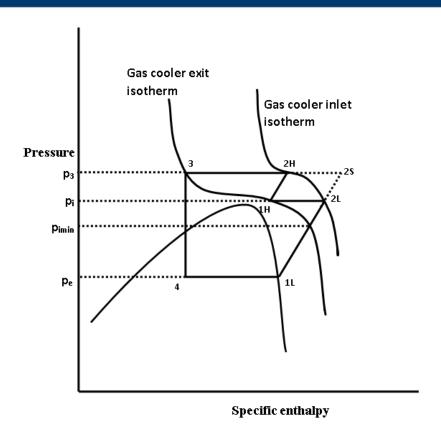


- > Most of the problems are due to high temperatures and high pressures together
- > If the high temperatures can be managed to be at the same level as with other refrigeration systems some problems can be alleviated.
- > Two stage compression is worth exploring
- > Intercooling will be essential between lower and upper stages.

The following criteria have to be considered for determining the inter-stage pressure:

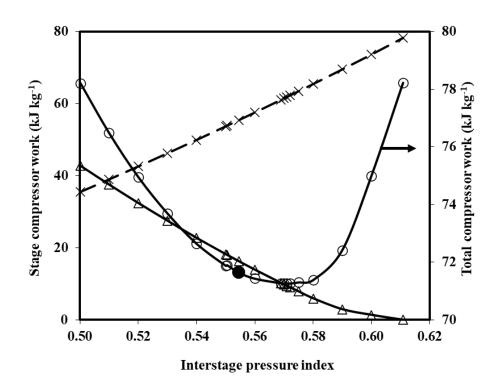
- i) The minimum exit temperature from the low stage should be at least the gas cooler exit temperature for any heat transfer to occur in the intercooler,
- ii) the exit temperature of the gas from low and high stage compressors should be equal,
- iii) the sum of work of compression in the two stages should be minimum, which is the conventional criterion and
- iv) the exergy loss during heat rejection (in the gas cooler plus the intercooler) should be the least.



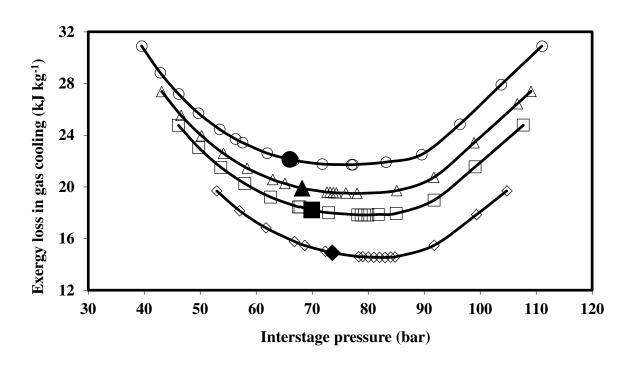


A schematic representation of CO2 transcritical two-stage compression refrigeration cycle on the p-h co-ordinate system

Interstage pressure index 
$$n = \frac{\ln p_i}{\ln(p_{1L}p_{2H})}$$







130
120
2 110
2 110
3 100
90
90
90
40
0.50
0.52
0.54
0.56
0.58
0.60
0.62
0.64
Interstage pressure index

Variation of exergy loss in gas cooling with inter-stage pressure for  $t_3$ =40 °C. Legend:  $\diamondsuit$ - $t_e$ =-10 °C,  $\square$ - $t_e$ =-20 °C, D- $t_e$ =-24.5 °C, O- $t_e$ =-30 °C. Darkened points correspond to the state of equality of exit temperatures of each compressor stage

Stage discharge temperatures for various inter-stage pressure indices for  $t_3$ =40 °C. Legend: High stage - open symbols, low stage - filled symbols.  $\diamondsuit$ - $t_e$ =-10 °C,  $\Box$ - $t_e$ =-20 °C, D- $t_e$ =-24.5 °C, O- $t_e$ =-30 °C



### Main advantage of 2-stage compression will be

- > Large volumetric efficiencies of each stage
- Lower leakage across the piston
- > Lower discharge temperatures from compressor- easy oil management

### **Conclusion**



### CO<sub>2</sub>

- > IS A NATURAL ENVIRONMENTALLY FRIENDLY WORKING FLUID
- > IT HAS EXCELLENT THRMODYNAMIC AND THERMOPHYSICAL PROPERTIES
- > IT HAS A WIDE RANGE OF OPERATION FOR POWER GENERATION AND REFRIGERATION
- > IT ENALES COMPACT SYSTEMS, REDUCING THE COST

### **CHALLENGES**

High pressure management.

## **Upcoming Technical Webinars**







Enabling Digital Transformation of Industry With AI and Big Data

Date and time: Thursday, 17 March 2022, 6:00pm - 7:00pm AWST

Voltage, Stability and Islanding Control in Microgrids

Date and time: Thursday, 31 March 2022, 3:00pm - 4:00pm AWST

Upcoming webinars: <a href="https://www.eit.edu.au/news-events/events/">https://www.eit.edu.au/news-events/events/</a>

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# Thank you for attending.

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